Workshop AFT - misura dei nitrati per acque potabili

Teorico: Misura in continuo "optical measurement principles"





Overview

Morning

- Definitions of important terms optical measurement
- Characteristics of the absorption of light (visible and invisible)
- Characteristics of nitrate and organic compounds

Afternoon

- Photometric construction: CAS51D Viomax
 - Nitrate sensor
 - SAC sensor
- Experiment: Turbidity and Absorption
- Measurement loop in drinking water
- Discussion



Definition of important terms optical measurement

- Dispersion and dissolved substances
- Absorption, color, intensity, scattering, turbidity
- Lambert Beer law





What do you see...? \rightarrow Flipchart





Dispersion and dissolved substances

Dispersion – heterogeneous fluid \rightarrow turbidity

- Emulsion: Two not mixable fluids
- Suspension: Solid in fluid
- Foam: Gas in Fluid

Dissolved substances – homogeneous solution \rightarrow color

Molecules, ions or atoms homogeneous distributed

Relevance for drinking water?



Inline Photometer

Color intensity at same concentration



Lambert-Beer-Law - Küvettenlänge





Different colors



Lambert-Beer-Law - Absorptionskoeffizient





Color intensity at same view points





Lambert-Beer-Law - Concentration





Summary Lambert-Beer-Law





SAC: specific absorption coefficient (definition)





What is Turbidity...?



What is Turbidity...?

• The reflection is a function of the size and shape of the particles





Summary

- Dissolved substances lead to absorption at certain wave lengths (<u>color</u>)
- Emulsions and air bubbles lead to scattering (<u>turbidity</u>)

Keep in mind:

"Every turbidity leads also to absorption – but NOT every absorption leads to turbidity!"



Characteristics of the absorption of light

- Visible and invisible light why do we see colors?
- Characteristics of nitrate and organic compounds (specific absorption)





Visible light: Colors and wavelengths



Characteristics of nitrate and organic compounds

- Nitrate: Strong absorbance at 214nm
 → Concentration = mg/L NO₃⁻
- Organic compounds: Strong absorbance* at 254nm \rightarrow SAC = m⁻¹

Are there any restrictions...?

*Each organic compound has its own ε – the correaltion to the concentration is only possible, if the matrix is stable (alcohols and suggars do not show any absorbance)

Lunchtime

....





Workshop Pratico: Misura in continuo

- Photometric construction: CAS51D Viomax
- Experiment: Turbidity and Absorption
- Measurement loop in
 - drinking water
- Discussion





Photometric construction: CAS51D Viomax





Photometric construction: CAS51D-AAA* (Nitrate)

$c = [mg/LNO_3^{-}] = A / d / \epsilon_{NO^{3-}}$





Photometric construction: CAS51D-AAC* (SAC)





Experiment – tubidity and absorption

"Every turbidity leads also to absorption – but NOT every absorption leads to turbidity!"



Online measurement of nitrate and SAC

Bypass (Panel-Solution)

- 1. Transmitter Liquiline CM44x
- 2. Sensor Viomax CAS51D
- 3. Flow chamber
- 4. Optional: Filter (1 & 0.5µm)
- 5. Optional: Flow monitoring
- 6. Optional: Connection for calibration









Examples



Discussion



Die Perspektive macht den Unterschied

Slide 28 06/20/2017 Stefan Vogel

